

Dehydrooligopeptides. XIV. Convenient Coupling of *N*-Carboxy- α -dehydroamino Acid Anhydride with Both Amine and Carboxyl Components¹⁾

Chung-gi SHIN,* Seiji HONDA, Katsuhiro MOROOKA, and Yasuchika YONEZAWA

Laboratory of Organic Chemistry, Faculty of Technology, Kanagawa University,
Rokkakubashi, Kanagawa-ku, Yokohama 221

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Synopsis. In order to further develop and expand the usefulness of *N*-carboxy- α -dehydroamino acid anhydride (Δ NCA), the optimum conditions of the acylation of Δ NCA with *N*-protected α -amino acids (AA) were thoroughly examined. Furthermore, various kinds of AA or dipeptides as a C-component were utilized in the acylation of Δ NCA, followed by condensation with an AA methyl ester as an *N*-component.

In connection with the application of *N*-carboxy- α -dehydroamino acid anhydride (Δ NCA; **1**),²⁾ we have already reported the syntheses of various kinds of dehydrooligopeptides (DHP).^{3–7)} At present, however, the isolation of *N*-protected AA- Δ NCA (**2**) from **1** and *N*-protected AA-OH (AA = α -amino acids) has been unsuccessful. Here, we wish to report the isolation of **2** in a pure form and new one-pot syntheses of a few DHP by the fragment condensation of **1** with both amine (N) and carboxyl (C) components.

Results and Discussion

In order to obtain **2** in a pure form by the acylation of **1** with *t*-butoxycarbonyl (Boc)-AA-OH, various procedures were examined extensively. Besides the condensing agent, the combination of the organic solvent and base were altered variously, as shown in Table 1. As a result, the acylation of equimolar *N*-carboxy-dehydrophenylalanine anhydride (Δ Phe-NCA; **1b**) with Boc-Phe-OH in CH₂Cl₂ in the presence of dicyclohexylcarbodiimide (DCC) and an organic base, such as pyridine, proceeded smoothly to give the expected Boc-Phe- Δ Phe-NCA (**2k**). In particular, as Tables 1 and 2 show, when **1b** and pyridine were used in a 2:1 ratio, the yield and the specific rotation of **2k** were found to be the highest (81% and -67.5° , respectively).

Table 1. Effect of the Catalyst and Solvent on the Yield of Boc-Phe- Δ Phe-NCA (**2k**)

Catalyst	Solvent	Time/h ^{a)}	Yield/%
Et ₃ N	CH ₂ Cl ₂	24	18.4
NMM ^{b)}	CH ₂ Cl ₂	24	31.3
DMAP ^{c)}	CH ₂ Cl ₂	10	67.6
Pyridine	CH ₂ Cl ₂	10	81.5
Pyridine	THF ^{d)}	12	62.3

a) At room temperature. b) *N*-Methylmorpholine.
c) 4-Dimethylaminopyridine. d) Tetrahydrofuran.

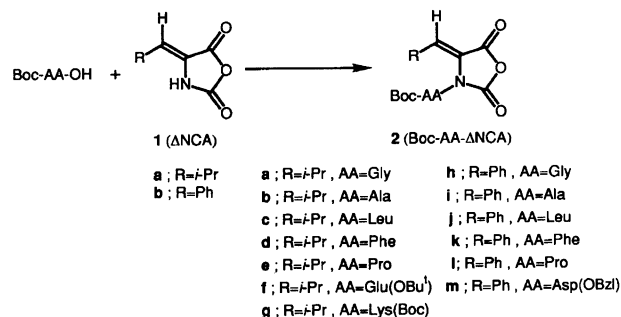
Table 2. Effect of the Molar Ratio of Pyridine to **1b** on the Specific Rotation of **2k**

Ratio of pyridine to 1b (mol/mol)	$[\alpha]_D^{25}/^\circ$ (c 1.0, MeOH)
5.0	-57.2
3.0	-61.8
1.0	-64.6
0.5	-67.5
0.25	-64.0

Furthermore, in the case of *N*-carboxy-dehydroleucine anhydride (Δ Leu-NCA; **1a**) and pyridine in a 1:1 ratio, it was found that the similar condensation of **1a** with Boc-Phe-OH also proceeded to give Boc-Phe- Δ Leu-NCA (**2d**) in the highest yield of 85.5%, according to Scheme 1. Consequently, similar *N*-acylations of **1a** and **1b** with various kinds of Boc-AA-OH were worked up to give the expected Boc-AA- Δ Phe-NCA (**2a–g**) as colorless needles and Boc-AA- Δ Leu-NCA (**2h–m**) as colorless syrups in 85 and 78% yields, respectively.

The yields, physical constants, and spectral data (IR and ¹H NMR) of **2a–g** and **2h–m** are summarized in Tables 3 and 4 respectively.

The structures of all of the new compounds **2** thus obtained were determined spectroscopically and gave satisfactory results in elemental analyses. In particular, in the IR spectra of **2**, the appearance of the characteristic strong absorption band in the 1803–1827 cm⁻¹ region due to the cyclic acid anhydride (–CO–O–CO–) clearly showed the maintenance of the Δ NCA ring system during the *N*-acylation. Furthermore, in the ¹H NMR spectra, the signals of the olefinic protons of **2a–g** appeared in the $\delta=6.68$ –6.15 region as a doublet



Scheme 1.

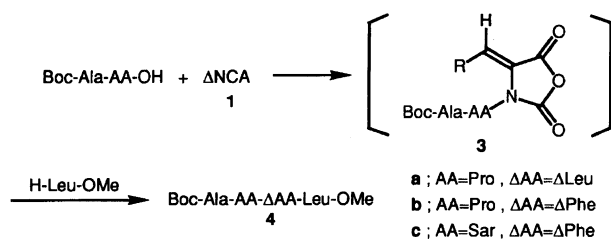
Table 3. The Yields and Melting Points of **2a—m**

Comd No.	Yield %	Mp ^{a)} $\theta_m/^\circ\text{C}$	Formula	Found (Calcd)/%		
				C	H	N
2a	79	Syrup	C ₁₄ H ₂₀ N ₂ O ₆	53.77 (53.84)	6.37 (6.45)	9.02 (8.97)
2b	92	Syrup	C ₁₅ H ₂₂ N ₂ O ₆	55.53 (55.20)	6.72 (6.80)	8.69 (8.58)
2c	93	Syrup	C ₁₈ H ₂₈ N ₂ O ₆	58.79 (58.68)	7.45 (7.66)	7.56 (7.60)
2d	86	Syrup	C ₂₁ H ₂₆ N ₂ O ₆	62.89 (62.67)	6.32 (6.51)	7.11 (6.96)
2e	94	Syrup	C ₁₇ H ₂₄ N ₂ O ₆	58.12 (57.94)	6.67 (6.87)	8.07 (7.95)
2f	77	Syrup	C ₂₁ H ₃₂ N ₂ O ₈	57.41 (57.26)	7.03 (7.32)	6.51 (6.36)
2g	71	Syrup	C ₂₃ H ₃₇ N ₃ O ₈	57.39 (57.13)	7.54 (7.71)	8.79 (8.69)
2h	79	121—122	C ₁₇ H ₁₈ N ₂ O ₆	59.11 (58.95)	5.17 (5.24)	8.18 (8.09)
2i	81	104—105	C ₁₈ H ₂₀ N ₂ O ₆	60.07 (59.99)	5.60 (5.77)	7.81 (7.77)
2j	80	127—128	C ₂₁ H ₂₆ N ₂ O ₆	62.88 (62.67)	6.35 (6.51)	7.08 (6.96)
2k	84	136—137	C ₂₄ H ₂₄ N ₂ O ₆	66.21 (66.04)	5.32 (5.54)	6.66 (6.42)
2l	75	90—91	C ₂₀ H ₂₂ N ₂ O ₆	62.37 (62.16)	5.54 (5.74)	7.34 (7.25)
2m	73	114—116	C ₂₆ H ₂₆ N ₂ O ₈	63.32 (63.15)	5.41 (5.30)	5.77 (5.67)

a) Colorless viscous syrups and colorless needles from hexane-ethyl acetate.

($J=10.0$ Hz) and those of **2h—m** in the $\delta=7.50$ — 7.13 region as a singlet. Based on the $^1\text{H NMR}$ data, the *Z*-configurational structures were also found to remain during the *N*-acylation.²⁾

In addition, the C-component Boc-alanylproline (Ala-Pro) was chosen for the similar *N*-acylation of **1**, according to Scheme 2. Although the isolation of Boc-Ala-Pro- $\Delta\text{Leu-NCA}$ (**3a**) and - $\Delta\text{Phe-NCA}$ (**3b**) in a pure form was unsuccessful, the desired Δ^3 -dehydro-tetrapeptides,^{8,9)} Boc-Ala-Pro- $\Delta\text{Leu-Leu-OMe}$ (**4a**) and Boc-Ala-Pro- $\Delta\text{Phe-Leu-OMe}$ (**4b**), were obtained in 25.3 and 39.0% yields, respectively, by immediate treatment of **3a** and **3b** with H-Leu-OMe as the *N*-component. Thus, the one-pot syntheses of **4** were successful, although the yields were considerably low. In order to increase the yield of **4**, suitable combinations of



Scheme 2.

Table 4. The IR, $^1\text{H NMR}$, and Specific Rotations of **2a—m**

Compd No.	IR, ν/cm^{-1} in KBr		$^1\text{H NMR}$, $\delta^a)$ $-\text{CH}= (J/\text{Hz})$	$[\alpha]_D^{25}/^\circ$ (c 1.0, MeOH)
	O=C-O	NHCO		
2a	1815	1716 1515	6.68d (10.0)	—
2b	1812	1707 1518	6.15d (10.0)	−37.7
2c	1812	1716 1515	6.50d (10.0)	−30.4
2d	1809	1713 1503	6.40d (10.0)	−15.1
2e	1812	1707	6.44d (10.0)	−34.0
2f	1812	1725 1518	6.44d (10.0)	−27.7
2g	1812	1707 1524	6.44d (10.0)	−15.2
2h	1803	1683 1521	7.18s	—
2i	1803	1686 1521	7.13s	−130.1
2j	1803	1680 1518	7.16s	−91.4
2k	1806	1695 1525	7.1–7.9m	−67.2
2l	1810	1692	7.15s	−67.2
2m	1827	1722 1521	7.14s	−27.3

a) Measured in CDCl_3 .

the basic catalyst and the solvent, and optimum ratios of the amount of the catalyst to the substrate **1** were studied under various conditions. It was found that 4-dimethylaminopyridine (DMAP) was superior to pyridine and the optimum molar ratio of DMAP to **1** was 1:1. Consequently, the compounds **4a** and **4b** were obtained in the highest yields of 68 and 78%, respectively (Table 5).

Furthermore, in place of Boc-Ala-Pro-OH, a similar work up of **1b** with consecutive Boc-alanylsarcocine (Ala-Sar) and H-Leu-OMe gave Boc-Ala-Sar- $\Delta\text{Phe-Leu-OMe}$ (**4c**) in a 79% yield.

Experimental

Melting points were determined with a Yamato (Model Mp-21) micro melting-point apparatus, and were not corrected. IR spectra were recorded with a Hitachi EPI-G2 grating spectrometer. $^1\text{H NMR}$ spectra were measured with a JEOL JMN PS-100 spectrometer in a CDCl_3 solution with tetramethylsilane as the internal standard. The specific rotations were measured in a 0.5-dm tube using a JASCO DIP-4 polarimeter (Japan Spectroscopic Co., Ltd.).

Boc-AA- ΔNCA (2**).** To a solution of an appropriate Boc-AA-OH (6.5 mmol) in CH_2Cl_2 (30 ml) was added DCC (7.7 mmol), with stirring, at -10°C for 20 min followed by addition of **1** (6.5 mmol) and pyridine (7.2 mmol). After stirring for 1 h at -10°C and at room temperature for 12

Table 5. The Yields and Physical Constants of **4a—c**

Compd No.	Yield %	Mp ^{a)} $\theta_m/^\circ\text{C}$	Formula	Found (Calcd)/%			$[\alpha]_D^{25}/^\circ$ (c 1.0) ^{b)}
				C	H	N	
4a	68	167—168	C ₂₆ H ₄₄ N ₄ O ₇	59.14 (59.52)	8.24 8.45	10.33 10.68	−25.3
4b	78	203—204	C ₂₉ H ₄₂ N ₄ O ₇	62.78 (62.34)	7.72 7.58	10.28 10.03	−89.1
4c	79	108—109	C ₂₇ H ₄₀ N ₄ O ₇	60.45 (60.88)	7.47 7.57	10.37 10.52	−21.6

a) Colorless needles from hexane–ethyl acetate. b) Measured in MeOH.

h, the solution was concentrated in vacuo. The residue was dissolved in a small amount of ethyl acetate and the precipitates were filtered off. The filtrate was diluted with ethyl acetate (300 ml), washed successively with 10% citric acid and brine, and then dried over anhydrous Na₂SO₄. Concentration in vacuo gave crystals of syrups, which were purified by recrystallization from hexane–ethyl acetate or on a silica-gel column using benzene–acetone (4:1 v/v) as the eluent to give **2** as colorless needles or syrups (Tables 3 and 4).

Boc-Ala-Pro-OH. To a solution of HCl·H-Pro-OMe (58 mmol) in CH₂Cl₂ (100 ml) was added Et₃N (58 mmol), with stirring, below 0 °C for 10 min. Boc-Ala-OH (53 mmol) and WSC·HCl (1-ethyl-3-(3-dimethylaminopropyl)-carbodiimide hydrochloride) (58 mmol) were added, with stirring, to the solution prepared above over 1 h. The solution was then stirred at room temperature for 12 h. After concentrating in vacuo, the residue was dissolved in ethyl acetate (300 ml) and washed successively with water, twice with 1 M HCl (140 ml) (1 M = 1 mol dm^{−3}), water, saturated NaHCO₃ aqueous solution (70 ml), and brine (70 ml), and finally dried over anhydrous Na₂SO₄. After removal of the solvent in vacuo, crude Boc-Ala-Pro-OMe was obtained in an ca. 80% yield. Subsequently, to a solution of the obtained dipeptide ester (33 mmol) in MeOH (100 ml) was added 1 M NaOH (36 ml) under cooling and the resulting solution was stirred at room temperature for 2 h. After concentrating, the residue was dissolved in water (300 ml) and the aqueous solution was washed with ethyl acetate (100 ml) and then made acidic to pH 2 with 6 M HCl and extracted three times with ethyl acetate (100 ml). The combined extracts were washed with brine and then dried over anhydrous Na₂SO₄. Concentration in vacuo gave crystals, which were recrystallized from hexane–ethyl acetate to give Boc-Ala-Pro-OH as colorless needles. Yield 67%, mp 147—148 °C. $[\alpha]_D^{25}$ −96.3° (c 1.0, MeOH). Found: C, 54.84; H, 7.84; N, 9.48%. Calcd for C₁₃H₂₂N₂O₅: C, 54.53; H, 7.75; N, 9.78%.

Boc-Ala-Sar-OH. In a similar manner, the coupling of Boc-Ala-OH with H-Sar-OMe and then ester hydrolysis of Boc-Ala-Sar-OMe were carried out to give a syrup, which was purified on a silica-gel column using CHCl₃–MeOH (10:1 v/v) as the eluent to give Boc-Ala-Sar-OH as a colorless viscous syrup. Yield 80%. $[\alpha]_D^{25}$ −33.8° (c 0.9,

MeOH). Found: C, 50.87; H, 7.67; N, 10.66%. Calcd for C₁₁H₂₀N₂O₅: C, 50.76; H, 7.75; N, 10.76%.

Boc-Ala-AA-ΔLeu-Leu-OMe (4a—c). After adding DCC (4.2 mmol) to a solution of Boc-Ala-AA-OH (3.5 mmol) in CH₂Cl₂ (10 ml), with stirring, at 0 °C for 30 min, the resulting solution was further treated with an equimolar amount of **1b** (3.4 mmol) and DMAP for 1 h and at room temperature for 14 h. The solution was then treated with a solution of H-Leu-OMe (3.9 mmol) in CH₂Cl₂ (10 ml) and the pH was adjusted to 8 with *N*-methylmorpholine (NMM), with stirring, for 12 h. After concentration in vacuo, the residue was dissolved in a small amount of ethyl acetate. After stirring at −10 °C for 30 min and removing the insoluble material, the filtrate was diluted with ethyl acetate (300 ml) and washed successively three times with 10% citric acid (30 ml), brine (10 ml), and saturated NaHCO₃ aqueous solution (10 ml), and then dried over anhydrous Na₂SO₄. Concentration in vacuo gave crystals, which were recrystallized from hexane–ethyl acetate to give **4a—c** as colorless needles (Table 5).

References

- 1) Part XIII: C. Shin, N. Takahashi, and M. Seki, *Bull. Chem. Soc. Jpn.*, **64**, 3575 (1991).
- 2) C. Shin, Y. Yonezawa, and J. Yoshimura, *Chem. Lett.*, **1981**, 1635.
- 3) Y. Yonezawa, T. Yamada, and C. Shin, *Chem. Lett.*, **1982**, 1567.
- 4) C. Shin, T. Yamada, and Y. Yonezawa, *Tetrahedron Lett.*, **24**, 2175 (1983).
- 5) C. Shin, Y. Yonezawa, and T. Yamada, *Chem. Pharm. Bull.*, **32**, 3934 (1984).
- 6) C. Shin and Y. Yonezawa, *Chem. Lett.*, **1985**, 519.
- 7) C. Shin, M. Ikeda, and Y. Yonezawa, *Agric. Biol. Chem.*, **49**, 2243 (1985).
- 8) C. Shin, Y. Yonezawa, and M. Ikeda, *Bull. Chem. Soc. Jpn.*, **59**, 3573 (1986).
- 9) In this paper, the symbol Δ³ indicates the position number of the double bond of the DHA residue from the *N*-terminus in sequence.